

2010

Synthesis and Uses of Polycarbonate

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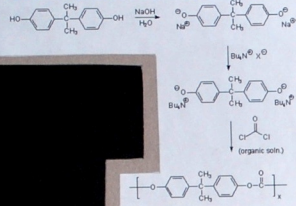
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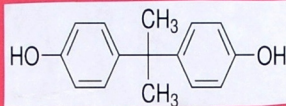
INTERFACIAL PROCESS

Synthesis of Polycarbonate through Interfacial Polymerization



4

Synthesis & Uses of Polycarbonate

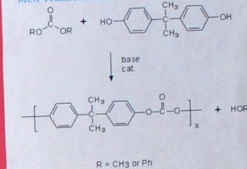


Bisphenol A

5

MELT TRANSESTERIFICATION

Melt Transesterification



4



Uses of polycarbonate

- equipment housings
- exterior automotive components
- outdoor lighting fixtures
- nameplates and bezels
- non automotive vehicle windows
- brackets and structural parts
- medical supply components
- plastic lenses for eyeglasses

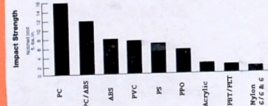
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Interfacial Process

Kinetically controlled reaction
Molecular weight via chainstopper
Amine catalysis in solvent
Low temperature
Washing and isolation is necessary
High molecular weight
Few side reactions
Requires phosgene and solvent

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2

Properties of Polycarbonate

- excellent physical properties
- excellent toughness
- very good heat resistance
- fair chemical resistance
- transparent
- moderate to high price
- fair processing

2

Conclusion

Polycarbonate is a very versatile material that can be used in many applications. It is used to make eyeglass lenses, water bottles, light fixtures, and windows. There are two main ways that polycarbonates are being synthesized commercially, interfacial and melt transesterification. Both processes synthesized polycarbonate from bisphenol A but both are very different. The interfacial process uses an amine catalyst, requires phosgene gas, and is kinetically controlled. The melt transesterification process uses a base catalyst, requires diphenyl carbonate, and is thermodynamically controlled. The interfacial process is generally used to synthesize polycarbonates that are used in bigger applications such as windows. The melt transesterification process is used to create polycarbonates that are light and used in eyeglass lenses.

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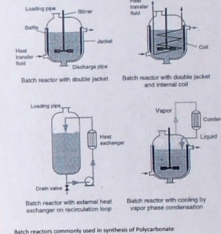
References

1. Swartz, Anionic, Synthesis and Thermal Properties of Polycarbonates Based on Bisphenol A by Single Phase Organic Solvent Polymerization. *Prog. Adv. Chem.* 2006, 13, 181-199.
2. Polymer Technology & Services. A Guide to Polycarbonate in General. http://www.pst.com/tech/pst_001.htm (accessed Nov 23, 2010).
3. Polymer Technology. Basic of Polymer Flow. <http://pdfdrive.net/flow/2008/05/04/index.html> (accessed Nov 23, 2010).
4. Chem4U. Synthetic Polymer Chemistry: Polycarbonate Synthesis. <http://chem4u.com/notebook/edu/chem4u/24/polym.htm> (accessed Nov 23, 2010).
5. Wikipedia. Bisphenol A. http://en.wikipedia.org/wiki/Bisphenol_A (accessed Nov 23, 2010).

Melt Process

Thermodynamic control
Molecular weight via extent of reaction
Base catalysis of condensation
High temperature
Direct isolation via extrusion
Low molecular weight
Side reaction may occur at high temperature
Requires diphenyl carbonate and basic catalyst
Does not require phosgene

1



Batch reactors commonly used in synthesis of Polycarbonate

3