

2018

Solid Electrolytes for Better Batteries: A Report on Current Research

Jonathan D. Young
Parkland College

Recommended Citation

This slide presentation for the Natural Sciences Poster Session at Parkland College summarizes research on solid electrolyte lithium batteries which hold an advantage over current use lithium-ion batteries.

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Solid Electrolytes for Better Batteries: A Report on Current Research

Jonathan Young

Parkland College

CHE 101-006

Dr. Nicely

Fall 2018



Image Source: <https://www.smithsonianmag.com/smart-news/researchers-figured-out-how-stop-lithium-batteries-spontaneously-combusting-180955641/>

Lithium-ion Battery Chemistry

- “A battery is a galvanic cell, or series of galvanic cells, that can be used as a portable, self-contained source of direct electric current.”¹
- Lithium-ion battery advantages vs. other chemistries¹:
 - Large cell potential
 - Low mass
 - Hundreds of charge cycles
- Wet cells (liquid electrolyte)²

The Holy Grail: Solid-Electrolyte Lithium Batteries

- “Solid-state batteries are less likely to catch fire than lithium-ion batteries”²
- Shorter charging time²
- “Could allow companies to squeeze more battery cells into the same size pack”²

Research into New Solid Electrolytes

- Heri, J.; Syahrial, A.Z.; Sudaryanto.; Kartini, E. Synthesis and Electrochemical Characterization of New $\text{Li}_2\text{O}-\text{P}_2\text{O}_5$ Compounds for Solid Electrolytes. *Int. J. Technol.* [Online], **2017**, 8, 1516-1524,
https://www.researchgate.net/publication/322081690_Synthesis_and_Electrochemical_Characterization_of_New_Li2O-P2O5_Compounds_for_Solid_Electrolytes (accessed Oct 16, 2018).
- “In this study, new compositions of the $x\text{Li}_2\text{O}-\text{P}_2\text{O}_5$ compounds, where $1 \leq x \leq 2$, were prepared through solid-state reactions.”³
 - Nonstoichiometric compounds: “any solid chemical compound in which the numbers of atoms of the elements present cannot be expressed as a ratio of small whole numbers”⁴

Research into New Solid Electrolytes: Synthesis Process

- Lithium carbonate (Li_2CO_3) and ammonium dihydrogen phosphate ($\text{NH}_4\text{H}_2\text{PO}_4$)³
 - “mixed using magnetic stirrers for two hours”³
 - ground in ceramic crucible with agate mortar for one hour³
 - Heated to 673 K for one hour to release H_2O , NH_3 , CO_2 molecules³
 - Slowly heated to 923 K for four hours³
 - “quenched in demineralized water and smoothed in an agate mortar for one hour”³

Research into New Solid Electrolytes: Characterization Techniques

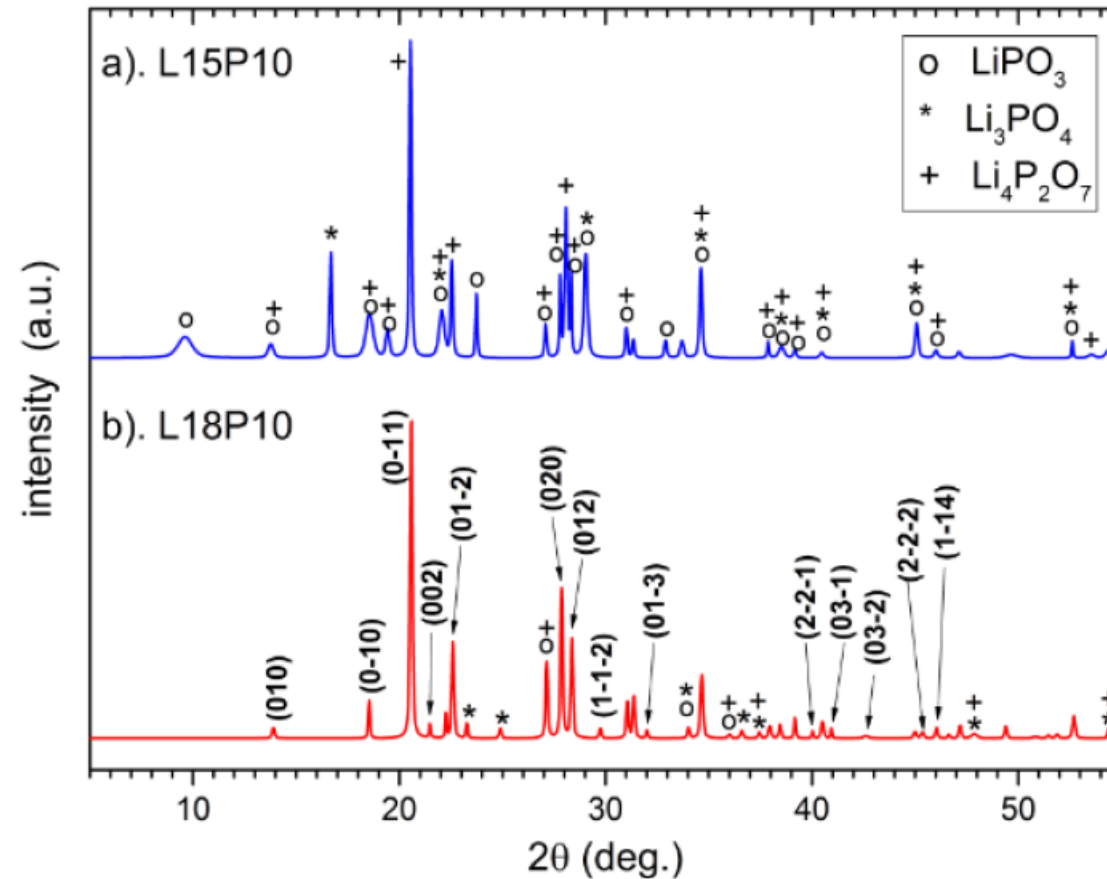
Techniques used by the researchers³:

- X-ray Diffraction Spectrometry
- Scanning Electron Microscopy
- Electrochemical Impedance Spectroscopy

Research into New Solid Electrolytes: X-ray Diffraction Spectrometry

- “used to identify crystals which are present in a mixture”⁵
- Sample placed in front of an x-ray beam, which is diffracted by the crystals as it passes through⁵
- Used to identify elements in sample and their proportions⁵

Research into New Solid Electrolytes: X-ray Diffraction Spectrometry

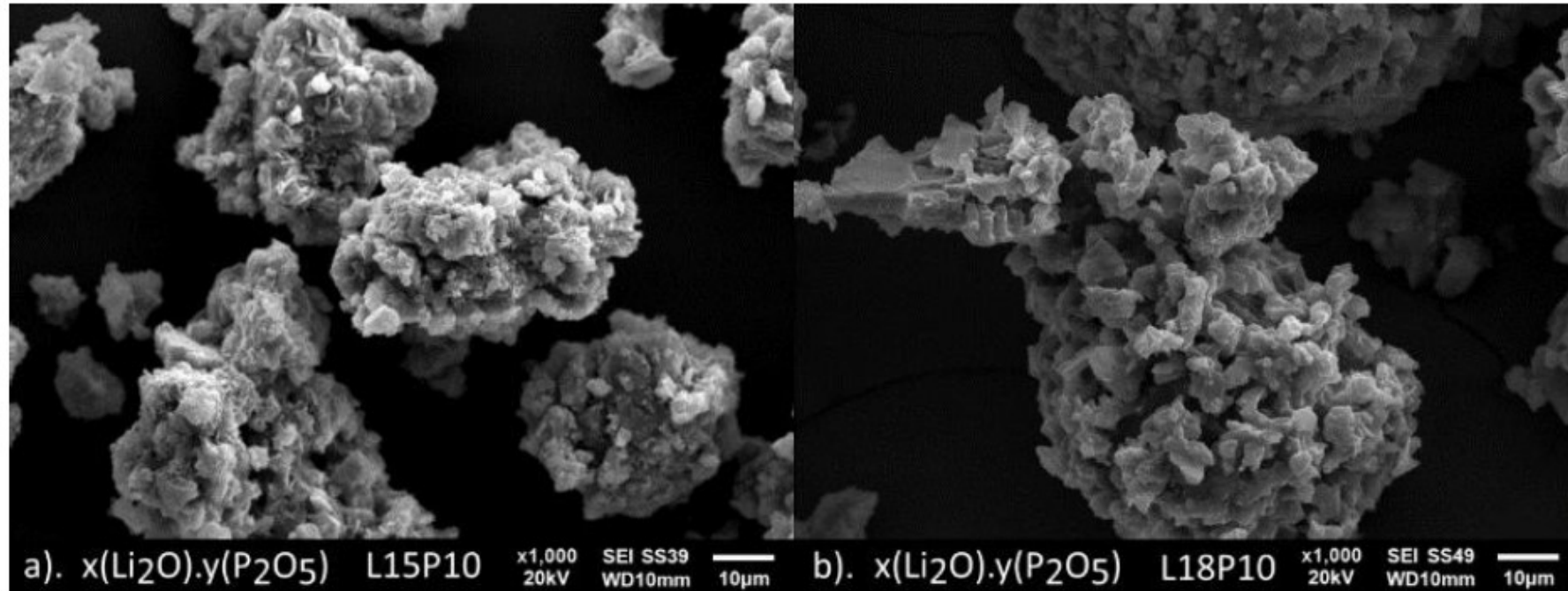


XRD pattern of $x\text{Li}_2\text{O}\text{-P}_2\text{O}_5$ compounds³

Research into New Solid Electrolytes: Scanning Electron Microscopy

- “uses a focused beam of high-energy electrons to generate a variety of signals at the surface of solid specimens”⁶
- reveals chemical composition, structure and orientation of crystals, morphology/texture⁶
- Images down to 5 microns (5×10^{-6} m)⁶

Research into New Solid Electrolytes: Scanning Electron Microscopy

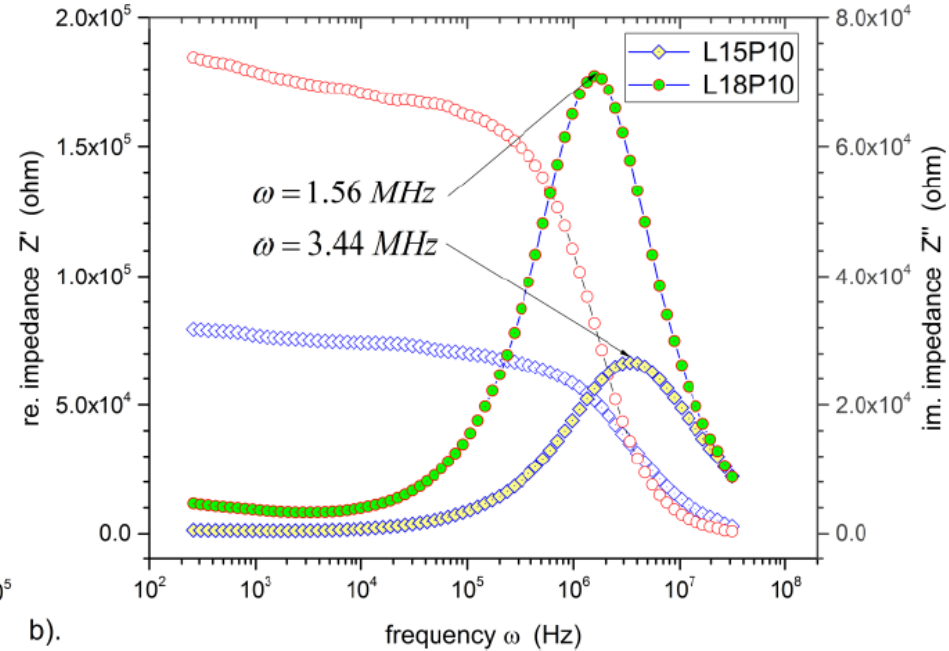
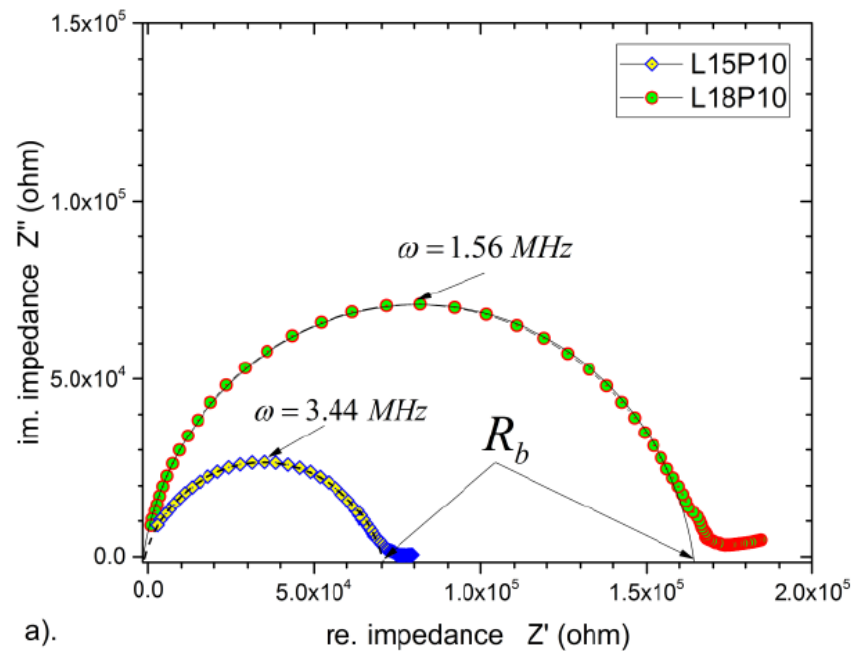


SEM micrograph of $x\text{Li}_2\text{O}-\text{P}_2\text{O}_5$ compounds³

Research into New Solid Electrolytes: Electrochemical Impedance Spectroscopy

- Impedance: Total opposition of an electrical component or circuit to current flow⁷
- EIS “characterize[s] a chemical process in terms of electrical measurements”⁸
- Response of chemical system to electrical stimulus reveals properties of the system⁸

Research into New Solid Electrolytes: Electrochemical Impedance Spectroscopy



Plots of impedance and frequency dependence of impedance of $x\text{Li}_2\text{O}-\text{P}_2\text{O}_5$ compounds³

Research into New Solid Electrolytes: Characteristics of the New Compound

- Produced two compounds: $1.5\text{Li}_2\text{O-P}_2\text{O}_5$ and $1.8\text{Li}_2\text{O-P}_2\text{O}_5$ ³
- Similar structure in both compounds and similar impedance³
- $1.5\text{Li}_2\text{O-P}_2\text{O}_5$ had lower resistance and better conductivity³
- Conductivity of both compounds was 3 orders of magnitude higher than Li_3PO_4 ,³ a common electrolyte in lithium-ion batteries⁹

Where To From Here?

- “There are a few next-generation battery technologies we’re looking at, and the most promising is an all solid-state battery.” Takeshi Uchiyamada, Chairman, Toyota Motor Corporation¹⁰
- “The solid-state battery will mark a turning point for e-mobility.” Heinrich Axel, VW Group Research¹¹

Where To From Here?



Image Source: <https://www.tesla.com/model3>

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